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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.005 Å R factor = 0.042 wR factor = 0.115 Data-to-parameter ratio = 10.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(Benzhydrylsulfinyl)acetamide

The title compound, $C_{15}H_{14}NO_2S$, was prepared by the esterification of benzhydrylthioacetic acid, followed by amidation and oxidation. There are two independent molecules in the asymmetric unit.

Comment

Novel acetamides have been discovered to have useful pharmaceutical activity on the central nervous system (Louis, 1978). They may be prepared by reacting the corresponding ester or acid halide with the appropriately substituted amine. The structure of the title compound, (I), is reported here as an early result in our study of this new series of compounds.



In the crystal structure of (I), there are two independent molecules in the asymmetric unit (Fig. 1 and Table 1). The crystal structure is shown in Fig. 2. There is an intermolecular hydrogen bond between O1 and N1 with an $O1 \cdots N1$ distance of 2.96 Å.

Experimental

Thionyl chloride (2 ml) was added into a benzene solution of benzhydrylthioacetic acid (1.955 g, 7 mmol) at room temperature,



Figure 1

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved The molecular structure of the aymmetric unit of (I), with 50% probability displacement ellipsoids and the atom numbering scheme.

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The crystal structure of (I).

and the mixture was heated for 1 h. The resulting chloride was filtered off and dissolved in methanol (10 ml) and, after refluxing for 1 h, ammonia was bubbled in at a high flow rate for 1 h. The solvent was evaporated and the residue was dissolved in ethanol. After 3 d, crystals of (I) were obtained from an ethanol solution.

Crystal data

C₁₅H₁₅NO₂S $D_x = 1.311 \text{ Mg m}^{-3}$ $M_r=273.35$ Mo $K\alpha$ radiation Monoclinic, $P2_1/n$ Cell parameters from 16675 a = 20.961 (1) Åreflections $\theta=2.3{-}27.4^\circ$ b = 9.7061 (5) Åc = 20.894(1) Å $\mu = 0.23 \text{ mm}^{-1}$ T = 298 (1) K $\beta = 139.336(1)^{\circ}$ $V = 2770.0 (3) \text{ Å}^3$ Prism, colorless $0.29 \times 0.28 \times 0.20 \text{ mm}$ Z = 8

Data collection

Rigaku R-AXIS RAPID	6218 independent reflections
diffractometer	3764 reflections with $F^2 > 2\sigma$
ω scans	$R_{\rm int} = 0.027$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.4^{\circ}$
(ABSCOR; Higashi, 1995)	$h = -27 \rightarrow 27$
$T_{\min} = 0.842, \ T_{\max} = 0.955$	$k = -12 \rightarrow 12$
47104 measured reflections	$l = -27 \rightarrow 27$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.042$	$w = (4F_o^2) / [0.002F_o^2 + 1.6\sigma(F_o^2)]$
$wR(F^2) = 0.115$	$(\Delta/\sigma)_{\rm max} < 0.001$
S = 1.00	$\Delta \rho_{\rm max} = 0.41 \ {\rm e} \ {\rm \AA}^{-3}$
3776 reflections	$\Delta \rho_{\rm min} = -0.30 \text{ e} \text{ Å}^{-3}$
373 parameters	

Table 1

Selected geometric parameters (Å, °).

\$1-02	1.485 (3)	N1-C2	1.317 (5)
S1-C1	1.797 (2)	N2-C17	1.322 (5)
S1-C3	1.838 (2)	C1-C2	1.517 (3)
S2-O4	1.487 (3)	C3-C4	1.512 (3)
S2-C16	1.789 (2)	C3-C10	1.515 (5)
S2-C18	1.837 (2)	C16-C17	1.515 (3)
O1-C2	1.226 (4)	C18-C19	1.520 (3)
O3-C17	1.222 (3)	C18-C25	1.513 (4)
O2-S1-C1-C2	60.8 (3)	O4-S2-C16-C17	-65.0 (3)
C3-S1-C1-C2	169.8 (3)	C18-S2-C16-C17	-172.8(3)
O2-S1-C3-C4	-60.3(3)	O4-S2-C18-C19	65.3 (3)
O2-S1-C3-C10	172.7 (2)	O4-S2-C18-C25	-168.4(2)
C1-S1-C3-C4	-170.3(3)	C16-S2-C18-C19	174.4 (3)
C1-S1-C3-C10	62.8 (2)	C16-S2-C18-C25	-59.3 (2)

All the H atoms were placed in calculated positions and then allowed for as riding, with C-H and N-H distances of 0.95 Å.

Data collection: RAPID-AUTO (Rigaku, 1998); cell refinement: RAPID-AUTO; data reduction: CrystalStructure (Rigaku/MSC, 2003); program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: CRYSTALS (Watkin et al., 1996); ORTEP-3 (Version 1.06; Farrugia, 1997); software used to prepare material for publication: CrystalStructure.

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 $> 2\sigma(F^2)$

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